## St. Xaviers School

## Syllabus and Lesson Planner

Class	IX	Syllabus		
Subject	CHEMISTRY			
Ch. No.	Name of Chapter	Торіс	Month	Week
1	THE LANGUAGE OF CHEMISTRY	Symbol of an element; valency; formula of radicals & formula of compounds	March	1
		Balancing of simple chemival equations.	April	2
		Relative Atomic Masses & Relative Molecular Masses.	•	
		Calculation of Relative Molecular Mass & percentage composition of a compound.		
2	CHEMICAL CHANGES AND REACTIONS	Types of chemical changes: Direct combination;		
		Decomposition; Displacement; Double decomposition.	May	2
		Energy changes in a chemical change: Exothermic & endothermic reactions with examples.		
3	ATOMIC STRUCTURE AND CHEMICAL BONDING	Structure of an atom, mass number and atomic	June	2
		number, Isotopes and Octet Rule		
		Electron distribution in the orbits - 2n2 rule. Reason for chemical activity of an atom.		
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		for chemical activity of an atom.		
		Electrovalent Bond: Atomic orbit structure for the formation of Electrovalent compounds ( e.g. NaCl,		
		MgCl2, CaO ).		
		Covalent Bond: Atomic orbit structure for the		
		formation of Covalent molecules on the basis of		
		duplet and octet of electrons.  Dobereiner's Triads, Newland law of Octaves,		
4	THE PERIODIC TABLE	Mendeleev's contributions; Modern Periodic Law, the	July	3
		Modern Periodic Table. (Groups and periods )	•	
		Special reference to Alkali metals, Alaline Earth metals, Halogens and Zero Group.		
5	WATER	Water as a universal solvent: Solutions as mixtures of	Aug/s	
		solids in water, saturated solutions	ept	2
		Qualitative effect of temperature on solubility.		
		Hydrated substances: Meaning & examples.		
		Anhydrous substances: Meaning & examples .		
		Properties: Efflorescence, Deliquescence, Hygroscopy Drying & Dehydrating agent - Meaning & examples		
		only.		
	CHAPTERS 1 to 5	1ST SEMESTER Revision	Aug/ sept	2
6	STUDY OF THE FIRST ELEMENT HYDROGEN	Position of the non-metal (Hydrogen) in the P. T.	Sept	3
		General group characteristics.		
		Preparation of hydrogen by general methods.		
		Preparation of hydrogen by laboratory method.		
		Preparation of hydrogen by Bosch process.		
		Oxidation & reduction reactions.		
7	STUDY OF GAS LAWS	Behaviour and characteristics properties of gases.	Oct	3
		Explanation in terms of molecular motion.		
		Boyle's Law & Charles Law, absolute zero.		
		Gas equation, standard temperature & pressure.		
		Relationship between Kelvin & Celsius scale.	Nove	
8	Atmospheric Pollution	Acid rain - composition, cause & its impact.	Nov	2
		Global warming. Ozone depletion.		
		Ozone depietion.		